

OPTIMIZATION OF LABETALOL HCL FILM BY USING QBD

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ABSTRACT

The present study was undertaken to develop and optimize Labetalol hydrochloride (HCl) loaded mouth dissolving film (MDF) using a Quality by Design (QbD) approach. Labetalol HCl is an antihypertensive drug with rapid gastrointestinal absorption but limited oral bioavailability due to extensive first-pass metabolism. Mouth dissolving films offer a convenient oral dosage form that rapidly disintegrates in the oral cavity and can improve ease of administration and patient compliance. In the present investigation, the QbD framework was applied by defining the Quality Target Product Profile (QTPP), identifying critical quality attributes (CQAs), performing risk assessment, and selecting critical formulation variables. A solvent casting method was selected for fabrication of the Labetalol HCl loaded MDF. Hydroxypropyl methylcellulose E15 (HPMC E15), polyethylene glycol 4000 (PEG 4000), and sodium lauryl sulfate (SLS) were selected as independent formulation variables. A Box-Behnken Design (BBD) was employed to investigate the influence of these variables on film thickness, folding endurance, and *in-vitro* drug release. Fifteen experimental batches (MQA301–MQA315) were prepared and evaluated, followed by preparation of the optimized formulation, MQA316. The optimized formulation consisted of HPMC E15 (331.947 mg), PEG 4000 (290.000 mg), and SLS (163.748 mg). The software-predicted values for thickness, folding endurance, and drug release were 100.44 μm , 210 folds, and 96.36%, respectively. The observed values for the optimized MQA316 formulation were 99.2 ± 0.4 μm thickness, 209 ± 0.816 folds folding endurance, and 95.90% drug release. The formulation exhibited a disintegration time of 27 ± 0.81 seconds, drug content of 99.17%, surface pH of 6.75 ± 0.07 , and acceptable weight variation. SEM analysis revealed a smooth film surface with few small pores. DSC and XRD studies indicated a reduction in drug crystallinity and conversion toward an amorphous state, while FTIR analysis indicated compatibility between the drug and excipients. Accelerated stability testing at $40 \pm 2^\circ\text{C}/75 \pm 5\%$ RH for three months showed no significant deterioration in the evaluated quality attributes. The close agreement between predicted and observed responses demonstrated the suitability of the QbD-based BBD approach for optimization of Labetalol HCl MDF.

Keywords: Labetalol hydrochloride; Mouth dissolving film; Quality by Design; Box-Behnken Design; Design of Experiments; HPMC E15; PEG 4000; Sodium lauryl sulfate.

1. INTRODUCTION

Quality by design (QbD) is a methodology used to build quality into products, by design. The QbD approach is used to ensure the pharmaceutical development is conducted in arrange to have in the end a systematic understanding of how process parameters affect a product (Silva *et al.*, 2017). According to ICH (International Conference Harmonization) Q8, QbD can be defined as “a systematic approach towards development that begins with predefined

objectives and emphasizes product and process understanding and process control, based on sound science and quality risk management” (**Patil and Pethe, 2013**). A very common tool or element used in the QbD is the quality target product profile (QTPP), which can be defined as “a dynamic product description that summarizes the quality characteristics expected to guarantee the product performance, stability, safety, and efficiency”. Mainly, QTPP includes the critical quality attributes (CQAs) and critical process parameters (CPPs). CQAs might be continued as attributes that characteristic of product quality and CPPs refers to critical process parameters or variables that can impact these characteristics. Therefore, the mix of the CQAs and CPPs allows the explanation of design space. The preparation of FDBF and QTPP may be organized based on earlier reported studies considering the essential requirements for manufacturing as well as techniques available for its characterization to get the desired film properties. A prepared FDBF should be handled without being harmed, ought to be physically stable and give a simple and pleasant administration. These properties may be translated into product quality attributes, such as appropriate organoleptic and mechanical properties (**Borges *et al.*, 2016**).

Transdermal drug delivery systems have gained considerable attention as an alternative to conventional oral administration because they can provide controlled and sustained drug release, avoid gastrointestinal degradation, reduce hepatic first-pass metabolism, and potentially improve patient compliance (**Alkilani *et al.*, 2022**). Among various transdermal systems, polymeric films are attractive because of their flexibility, ease of application, relatively simple manufacturing process, and ability to incorporate drug with suitable penetration-enhancing and film-forming agents. The development of an optimized Labetalol HCl transdermal film may therefore provide a promising approach for achieving prolonged drug release and maintaining more consistent drug concentrations (**Kotnala *et al.*, 2025**).

Labetalol hydrochloride, 2-Hydroxy-5-[1-hydroxy-2-[(1-methyl-3-phenylpropyl) amino] ethyl]-benzamide, a non-selective α , β -adrenoceptor antagonist which is used in the treatment of hypertension (**Jayanthi, 2018**). It is appreciably soluble in lower and higher pH solutions, with minimum solubility between pH 6 to 10. The drug shows variable bioavailability ranging from 10-80% which may be attributed to its instability in alkaline pH and poor absorption due to precipitation. It is administered in doses ranging from 50-200 mg twice a day due to its shorter half life of 3-6 hrs suggesting the need for sustained release formulation (**Murakami *et al.*, 2021**).

Labetalol HCl is administered orally in conventional tablet dosage forms as well as by other routes when rapid blood-pressure control is required. Following oral administration, the drug is absorbed from the gastrointestinal tract; however, its systemic availability is influenced by extensive hepatic first-pass metabolism (**Abdullah and Yusof, 2019**). Consequently, only a fraction of the administered dose reaches systemic circulation in unchanged form. Variability in absorption and metabolism may contribute to variations in plasma drug concentrations and therapeutic response. In addition, conventional oral dosage forms require swallowing with or without water and may not always be convenient for patients who have difficulty swallowing tablets (**Bow and Blomme, 2026**).

In the present study, a QbD-based approach was employed for the development and optimization of Labetalol HCl polymeric film. Appropriate polymers and other formulation components were selected based on their functional properties, while critical formulation

variables were systematically investigated using Design of Experiments. The developed films were evaluated for physicochemical and mechanical properties, drug content, *in-vitro* drug release, and other relevant performance characteristics. The optimized formulation was subsequently identified based on predefined acceptance criteria for the CQAs. The study aims to demonstrate the applicability of the QbD approach in developing a robust Labetalol HCl film with desirable quality attributes, reproducible performance, and potential suitability for transdermal drug delivery.

2. MATERIALS AND METHODS

2.1 Chemicals

Labetalol hydrochloride were obtained from Swapnroop Drugs & Pharmaceuticals Aurangabad, India, a reputable supplier of analytical reagents. Loba chemie pvt.ltd. Mumbai, India provided the HPMC E15, SLS, Citric Acid, Vanillin, Xylitol, Propylparaben, Methanol, Potassium dihydrogen phosphate and Disodium hydrogen phosphate. Chemdyes Corporation pvt. Ltd. Gujarat, India provided the PEG 4000.

2.2 Preformulation Studies of Labetalol Hydrochloride

The preformulation studies of Labetalol hydrochloride was carried out through organoleptic and physicochemical characterization for the compatibility studies of drug and excipients.

2.2.1 Organoleptic Characteristics

Organoleptic characteristics such as color, texture, and odor of the drug substance were determined (McMahon, 2008).

2.3 Evaluation of Physicochemical Properties and Analytical Characteristics

Physicochemical properties such as melting point and solubility. Analytical characteristics such as UV-visible spectrophotometric analysis, DSC, XRD, and FT-IR studies were carried out for the quality evaluation and characterization of Labetalol hydrochloride.

2.3.1 Melting Point

The melting point of Labetalol hydrochloride was determined using a melting point apparatus.

2.3.2 Solubility Profile

The solubility of the drug was determined in distilled water, phosphate buffer pH 6.8, and methanol, as per standard protocol and pharmacopoeia (Segall, 2023).

2.3.3 UV Spectroscopy

➤ Detection of $\lambda_{\max}$

A standard stock solution of drug was prepared by dissolving accurately weighed 5 mg of drug in 5 ml of distilled water (Solution A, i.e., 1000 $\mu\text{g/ml}$) and then 1ml of this solution was diluted to 10ml with distilled water (Solution B, i.e., 100 $\mu\text{g/ml}$). From this solution, further dilutions of 5, 10, 15, 20, 25 $\mu\text{g/ml}$ were prepared, and $\lambda_{\max}$ was determined by using scanning mode of UV-visible spectrophotometer.

➤ Calibration Curve in Methanol

A standard stock solution of drug was prepared by dissolving accurately weighed 5 mg of drug in 5 ml of methanol (Solution A, i.e., 1000 $\mu\text{g/ml}$) and then 1ml of this solution was diluted to 10ml with methanol (Solution B, i.e., 100 $\mu\text{g/ml}$). Different aliquots of solution B were diluted up to 10 ml with methanol so as to get solutions of 5, 10, 15, 20, and 25 $\mu\text{g/ml}$ concentration. Absorbance of each of the above dilutions were measured at the detection wavelength.

➤ Calibration Curve in Phosphate Buffer pH 6.8

A standard stock solution of drug was prepared by dissolving accurately weighed 5 mg of drug in 5 ml of phosphate buffer pH 6.8 (Solution A, i.e., 1000 µg/ml) and then 1ml of this solution was diluted to 10ml with phosphate buffer pH 6.8 (Solution B, i.e., 100 µg/ml). Different aliquots of solution B were diluted up to 10 ml with phosphate buffer pH 6.8 so as to get solutions of 5, 10, 15, 20, and 25 µg/ml concentration. Absorbance of each of the above dilution was measured at the detection wavelength (Zinjad *et al.*, 2019).

2.3.4 X-Ray Diffraction (XRD)

Crystallinity of Labetalol hydrochloride was evaluated by using X-ray diffractometer, using Nifiltered Cu-K α radiation source. The tube voltage of 40 kV, tube current 30 mA and scanning rate 5°C/min over a range of 5-40° of 2 θ diffraction angle at room temperature, patterns were formed with a step width of 0.04°C.

2.3.5 Differential Scanning Calorimetry (DSC)

Thermal analysis of Labetalol hydrochloride, was carried out using differential scanning calorimetric method. Samples equivalent to approximately 3.7 mg Labetalol hydrochloride was placed in thermetically sealed aluminium pans and heated over the temperature range of 25°C-250°C at a constant rate of 40°C/min under a nitrogen purge (20 ml/min).

2.3.6 FTIR Spectroscopy

The FTIR spectrophotometer was used to record the FTIR spectra of Labetalol hydrochloride at room temperature in the 4000-400 cm⁻¹ ranges. The sample was made by grinding it with dry KBr powder and compressing the powder in a hydraulic press (Sládková *et al.*, 2017).

2.4 Selection of Process for the Fabrication of MDF

There were various methods that could be used for the fabrication of MDF. Among those, the solvent casting method was selected as it was easy and most suitable for Labetalol hydrochloride loaded MDF formulation on the basis of literature survey, drug-excipient characteristics and availability of materials. In this method, a water-soluble polymer and plasticizer were dissolved in water, while the remaining excipients were separately dissolved. These solutions were then carefully mixed together with continuous stirring at 40°C and 300 rpm. During this blending process, the drug, dissolved in a suitable solvent, was slowly added while maintaining constant stirring. A vacuum was employed to remove any trapped air. The resulting solution was poured into molds, allowed to dry, and subsequently cut into the desired sizes and shapes (Singh *et al.*, 2012).

2.5 Design and Development of MDF Using QbD Approach

Steps involve in QbD approach were shown below:

Step 1: To establish a quality target product profile (QTPP) for the MDF.

Step 2: Determination of the critical quality attributes (CQAs) of the MDF.

Step 3: Incorporated a risk assessment exercise to identify the Critical Process Parameters (CPPs).

Step 4: Involved drafting an informal design of experiment (DoE) to determine optimal settings.

2.5.1 Defining the Quality Target Product Profile (QTPP)

The QTPP refers to a set of predefined characteristics and attributes that the final product was intended to have in order to meet specific quality and performance criteria. The QTPP criteria

for the development of MDF was identified using the results of the literature study and an awareness of the characteristics of the product and process (Joshi *et al.*, 2012).

2.5.2 Identifying the Critical Quality Attributes (CQA)

The CQAs indicate characteristics that have a significant impact on product performance. It ought to be chosen taking into account QTPP requirements. The CQAs taken into account for the creation of MDF include thickness, surface pH, *In-vitro* disintegration time, folding endurance, *In-vitro* dissolution, tensile strength, etc (Shantilal *et al.*, 2021).

2.5.3 Identification of Critical Material Attributes (CMA) and Critical Process Parameters (CPP)

The third step in drug development based on QbD was determining CMAs and CPPs. The CMAs and CPPs were identified based on the literature study and their impact on the CQAs. Concentration of Polymer, Plasticizer, Surfactant, and Nature of Excipients were the CMA's optimized parameters. Stirring speed, stirring time, stirring temperature and drying time were the CPPs under investigation. For the purpose of achieving the target quality, CPPs were chosen depending on the process parameter whose variability has an effect on CQAs (Saxena *et al.*, 2023).

2.5.4 Risk Assessment Studies and Screening of Factors for MDF

Risk assessment studies were carried out to pinpoint CMAs or CPPs with a significant impact on the CQAs of MDF. Potential high-risk factors that could influence the quality of the final MDF formulation were catalogued using an Ishikawa fishbone diagram. Risk levels for both formulation and process parameters were documented in the risk estimation matrix (REM), with each attribute classified as low, medium, or high risk based on its influence on CQAs. Parameters assessed as high risk were further investigated using the DoE to establish the design space (Mehrotra and Kumar, 2023).

2.6 Optimization of Formulation

The formulation optimization was typically done by simply adjusting each variable separately. However, it is difficult to develop an ideal formulation since the technique provides no information about how the variables interact. Therefore, to reduce the number of runs and errors, a computer-aided optimization process was used and for that Design Expert 13.0.15 (Stat-Ease) was utilized (Sahoo *et al.*, 2022).

2.6.1 Selection of Experimental Design

There were various experimental designs including full factorial, central composite, boxbehnken, mixture and D-optimal were considered useful for the computer-aided optimization process of MDF. Among those, on the basis of literature survey the most appropriate design was chosen and applied to optimize the MDF formulation of Labetalol HCl (Patel *et al.*, 2023).

2.6.2 Selection of Independent variables and Dependent or Response variables

Based on the literature review independent variables were selected and the values of the independent variables were taken on the basis of hit and trial method and were set at low (-1) medium (0) and high (+1) levels. While response variables were selected on the basis of significant impact of independent variables on the fabrication and performance of the final MDF formulation (Liu *et al.*, 2021).

2.6.3 Fabrication and Characterization of Batches of MDF of Labetalol Hydrochloride Suggested by Selected Experimental Design

On the basis of numerous independent variables, the selected experimental design offered several batches of MDF having different combinations. The MDF batches were prepared and evaluated for selected dependent variables. Total fifteen batches were prepared on the basis of recommendation of software and labelled as MQA301-MQA315. The batches were evaluated to formulate an optimized batch MQA316 similarly.

2.6.4 Statistical Analysis for MDF Formulation

➤ Fit Summary

The Fit Summary compiles the crucial facts required to choose the ideal model's initial condition. "The Whitcomb Score" was used to select the suggested model or models. The proposed model ought to be regarded as an ideal spot to start for the model fitting (Soni *et al.*, 2021).

➤ Sequential Sum of Squares

Sequential sum of square analyses the response's variability so that the error can be reduced. A sequential sum of squares calculates the amount of variance that may be explained (increased regression sum of squares) or alternately, the amount of error that can be decreased (decreased error sum of squares) (Xu, 2017).

➤ Analysis of Variance (ANOVA)

The ANOVA studies were generated from the software to evaluate the p-values, F-ratio, and lack of fit. In order to assess how well the experimental data fit, the regression coefficient (R^2) and adjusted R^2 were calculated. The following regression equation was used for prediction of the responses,

The polynomial equation generated by this experimental design is given as,

$$R = \beta_0 + \beta_1 A + \beta_2 B + \beta_3 C + \beta_{12} AB + \beta_{13} AC + \beta_{23} BC + \beta_{11} A^2 + \beta_{22} B^2 + \beta_{33} C^2$$

Where R was the response and A, B, and C were the response factors and β_0 , β_1 , β_2 and β_3 were the linear coefficients of the factors. β_{12} , β_{13} and β_{23} were the interaction coefficients and β_{11} , β_{22} and β_{33} were the quadratic coefficients (Reji *et al.*, 2022).

➤ Response Surface Analysis for MDF Formulation

The relationship between independent and response variables was further elucidated by constructing 3D response surface graphs for each response variable and the effect of independent variables on each response variable was studied.

➤ 3D Response Surface

Three-dimensional (3D) response surface plots were utilized to understand how the elements studied in mulch interact with each other, influencing their responses and correlations.

➤ 2D Contour Plot

The two-dimensional (2D) contour plot was used to demonstrate the connection between the responses, mixture elements, and/or numerical factors.

➤ Perturbation Graph

The perturbation plot simplifies the process of comparing the influence of each factor at a particular point within the design space. When plotting the response, one element was varied across its range, while the remaining variables were held constant. The reference point was set by "Stat-Ease" to be the middle (coded 0) for all the factors by default.

➤ Diagnostic Analysis

Regression quality was evaluated using residual plots. To verify the accuracy of these assumptions and provide guidance on how to improve the model, various residual plot types can be utilised (Selvamuthu *et al.*, 2018).

➤ Normal Plot of Residuals

A normal plot of residuals, also known as a normal probability plot of standardized residuals, was a graphical tool used in statistics to assess whether the residuals from a regression or statistical model followed a normal distribution. Residuals were the differences between observed data points and the values predicted by a statistical model.

➤ Residuals vs Predicted

The residuals were plotted against the descending expected response values in this graph, it serves as a visual verification for the constant variance assumption.

Residuals vs Run

The plot was used to view a very helpful plot known as "Outlier t" that displays how many standard deviations a particular run deviates from what one would anticipate from all the others (Hussain *et al.*, 2022).

➤ Predicted vs Actual

A comparison graph between the observed (actual) and projected response levels. It aids in identifying observations that the model does not accurately predict.

2.7 Formulation of Optimized MDF Formulation of Labetalol Hydrochloride

In order to find out the optimized composition of Labetalol hydrochloride loaded MDF, the desired constraints for response variables were set and accordingly fed into the software. Considering the desired constraint, the software generates distinct formulations. The formulation with maximum desirability was fabricated

2.8 Characterization of Optimized Labetalol Hydrochloride Loaded MDF

2.8.1 Measurement of Thickness

The thickness of the formulation was determined using a digital vernier caliper (Figure 9). Measurements were taken at 5 distinct points on the film i.e., central and each of the four corners of the film. Subsequently, the mean thickness was calculated. For measurement of uniformity of thickness, 3 films were randomly selected from each batch, and their thickness was measured (Martin *et al.*, 2017).

2.8.2 In-vitro Drug Dissolution Study

An *In-vitro* dissolution study of the MDF was conducted using a USP dissolution apparatus type-I (basket type) as illustrated in the Figure 10. The dissolution medium consisted of 300 mL of phosphate buffer (pH 6.8) and was maintained at $37^{\circ}\text{C} \pm 0.5^{\circ}\text{C}$ with a rotation speed of 50 rpm. Each film was securely placed inside the basket. At specific time intervals (0.5, 1, 2, 4, 8, and 10 minutes), 5 mL aliquots were withdrawn from each vessel, filtered, and subsequently subjected to analysis for Labetalol HCl using a spectrophotometer (Shimadzu-1800) at 302 nm. To maintain the sink condition, the volume in each vessel was replenished with 5 mL of fresh buffer (Upreti *et al.*, 2014).

2.8.3 Estimation of Folding Endurance

The folding endurance of the MDF was assessed by repeatedly folding it at a specific point until it reached a point of breakage or showed signs of damage. The recorded value for folding endurance indicates the total number of complete folding cycles accomplished before

failure occurred. A greater folding endurance value suggests robust mechanical characteristics and increased resilience for the MDF.

2.8.4 Estimation of Drug Content Uniformity

Drug content uniformity was assessed to ensure the even distribution of the drug within the films. This involved dissolving each film completely by stirring it in a volumetric flask filled with 10 ml of 6.8 pH phosphate buffer. The filtered solution was then utilized for measuring the absorbance at 302 nm to estimate the drug concentration, employing a Shimadzu 1800 UV-Spectrophotometer (Zayed *et al.*, 2020).

2.8.5 Estimation of pH

The pH of the MDF was determined by using a calibrated pH meter. Initially, the film was dissolved in 5 ml of phosphate buffer (pH 6.8). Subsequently, the pH of this solution was measured using a calibrated pH meter. This method was employed to ensure precise measurement of the pH of the MDF, a critical factor in evaluating its suitability for the oral cavity and its potential influence on taste and comfort during usage.

2.8.6 Estimation of Weight Variation

The weight variation test validates the uniformity of the fabricated MDF. Three film samples, each with an area of 6 cm² (2 cm x 3 cm), were individually extracted from each batch. The weight of each sample was measured, and the standard deviation of these measurements was computed. The average weight of the film should not exceed a deviation of $\pm 7.5\%$ (Ouda *et al.*, 2020).

2.8.7 Estimation of Disintegration Time

It is the point at which film begins to breakdown when exposed to water. The film was placed in a Petri dish containing 25 mL of water, and the time after the film completely disintegrated or collapsed was recorded. 30 seconds or less is thought to be the ideal disintegration time for MDF.

2.8.8 SEM of Optimized MDF

SEM allows for high-resolution imaging, which can reveal the surface morphology of the MDF. This is important for understanding its texture, smoothness, presence of any irregularity, and uniform distribution of the drug throughout the film formulation.

2.8.9 DSC of Optimized MDF

Thermal characterization was evaluated using the DSC analysis. MDF was cut into fine pieces using a sharp cutter, placed in thermetically sealed aluminium pans, and heated over a temperature range of 25°C–250°C at a constant rate of 40°C/min under a nitrogen purge (20 ml/min) (Kukal, 2026).

2.8.10 FTIR of Optimized MDF

The FTIR spectrophotometer was used to record the FTIR spectra of Labetalol hydrochloride loaded MDF at room temperature in the 4000-400 cm⁻¹ range. The sample was made by grinding it with dry KBr powder and compressing the powders in a hydraulic press.

2.8.11 XRD of Optimized MDF

XRD was performed to study the crystalline behavior of the drug after formulating it into a final formulation. The MDF underwent XRD analysis using an X-ray powder diffractometer equipped with a conventional Cu sealed X-ray tube, operated at a voltage of 40 kV and a current of 30 mA. The XRD profiles were registered within the 2θ range, employing a scanning rate of 4°/min (Senthilkumar and Vijaya, 2015).

2.8.12 Stability Study of MDF

To ascertain the impact of temperature on MDF stability throughout the specified duration, a stability analysis of the optimized formulation was performed. The formulation, enclosed in between two aluminium layers, was subjected to three months of accelerated stability testing at temperature $40\pm 2^\circ\text{C}$ and $75\pm 5\%$ relative humidity in accordance with ICH guidelines. At certain intervals of time (1 month, 2 months and 3 months.), samples were taken out, and physical and chemical characteristics such as thickness, disintegration time, surface pH, drug content, weight variation, folding endurance and *in-vitro* drug release were examined (Majeed *et al.*, 2017).

3. RESULT AND DISCUSSION

3.1 Preformulation Studies of Labetalol Hydrochloride

3.1.1 Organoleptic Characteristics

Table 1: Organoleptic Characteristics of Labetalol Hydrochloride

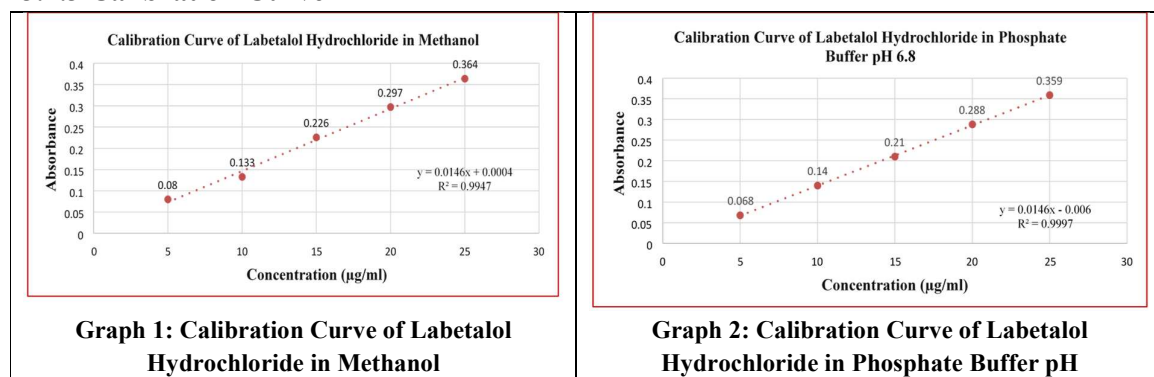
Properties	Observation
Color	White to off-white
Texture	Crystalline powder
Odor	Odorless
Taste	Slightly bitter

3.1.2 Melting Point

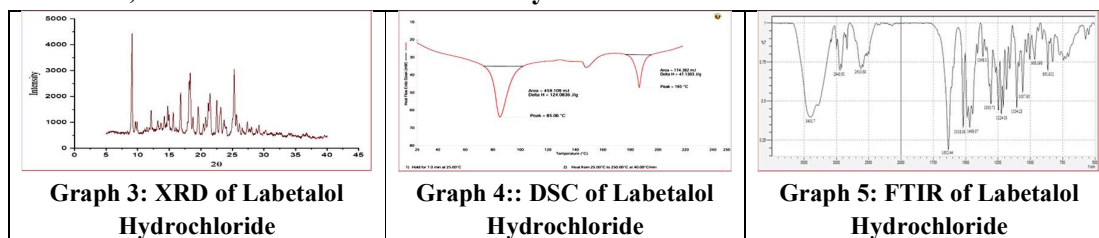
Table 2: Melting Point of Labetalol Hydrochloride

Melting Range		Average
Starting point	End point	
183°C	185°C	184°C
184°C	185°C	184.5°C
182.5°C	186°C	184.25°C

3.1.3 Calibration Curve



3.1.4 XRD, DSC and FTIR of Labetalol Hydrochloride



3.2 Process for the Fabrication of MDF

There were various methods for the fabrication of MDF as previously mentioned. Among those, the solvent casting method was selected as it was easy and most suitable for Labetalol hydrochloride loaded MDF formulation.

3.2 QbD Approach for MDF

3.2.1 Quality Target Product Profile for Labetalol Hydrochloride Loaded MDF

Table 3: Quality Target Product Profile for the Formulation of MDF

QTPP element	Target	Justification
Dosage form	Mouth dissolving film	This dosage form is suitable for paediatrics and geriatrics, as it is easy to ingest without water, making it ideal for patients with gastrointestinal issues. It's costeffective, portable, stable, and enhances patient compliance by reducing dosing frequency.
Dosage design	Immediate release film	Immediate release film plays a crucial role in ensuring timely and effective treatment of cardiovascular conditions, improving patient adherence, and addressing emergency medical needs.
Route of administration	Oral	Better patient compliance.
Dosage strength	5 mg Labetalol hydrochloride	It directly affects the efficacy of the drug, making it essential for achieving the desired therapeutic outcome. Insufficient dosage may lead to inadequate heart rate reduction, while excessive dosage can result in adverse effects, such as bradycardia.
Stability	At least 12 months at room temperature	Maintain quality, safety and efficacy throughout product life cycle.
Drug product quality attributes	Physical appearance • Surface texture • Transparency	Meeting the same compendia or other applicable quality standard
	Performance attributes • Surface pH • Moisture content	
	• <i>In-vitro</i> disintegration time	
	Mechanical properties • Folding endurance • Thickness • Tensile strength	
	Drug content • <i>In-vitro</i> dissolution	
	Identification • DSC • XRD • FTIR	

Table 4: CQA, CMA and CPP for the Formulation of MDF

CQA element		Target	Justification
Physical attributes	Appearance	Uniform surface texture	Doesn't have any effect on safety and toxicity.
Performance attributes	Thickness	5-200 μm	Thickness of film directly affect the Disintegration and dissolution. Hence, it should be optimum.
	Folding endurance	5-300 times	The target is set to ensure patient acceptability.
	Weight variation	SD $> \pm 7.5\%$	The target is set to ensure dose uniformity.
	Tensile strength	9-17g/mm ²	It gives an indication of strength of film. Hence, it should be moderate.
	Surface pH	6.2 to 7.6	The target is set to ensure patient acceptability.
	<i>In-vitro</i> disintegration time	5-30 sec	It has a direct effect on bioavailability. Hence, it should disintegrate within a minute.
	<i>In-vitro</i> dissolution	Maximum release in less than 10 Minutes	It directly affects the bioavailability. Hence, this attribute was studied throughout the finished dosage form.
	Content uniformity	85-115%	Content uniformity must be within limit to ensure uniform distribution of API.
Attributes		Effect on CQA	Control strategy
CMAs	Concentration of Polymer	Performance attributes	concentration range was selected from preliminary studies.
	Concentration of Plasticizer		
	Concentration of Surfactant		
	Nature of Excipients	Appearance	All excipients were water soluble.
CPPs	Stirring speed	Physical and performance attributes	300 RPM
	Stirring time	Homogeneity and performance attributes of final product	115-120 minutes
	Stirring temperature	Performance attributes and film homogeneity	39-40°C
	Drying time	Performance attributes and physical efficacy	6 to 9 hours

Table 5: REM for Initial Risk Assessment of Formulation Variables

Drug product CQAs	Formulation Variables		
	HPMC E15	PEG 4000	SLS
Folding endurance	High	High	Low
Thickness	High	Low	Low
<i>In-vitro</i> drug release	High	Medium	Medium

Table 6: Justification for Formulation Variables Inceptive Risk Assessment

Formulation Variables	Drug product CQAs	Risk Assessment	Justification
HPMC E15	Folding endurance	High	Folding endurance of MDF increases as the concentration of HPMC E15 increases. There was direct linear relationship. Hence, the risk was high.
	Thickness	High	Thickness increases as the concentration of HPMC E15 increases. There was direct linear relationship

			in between. Hence, the risk was high.
	<i>In-vitro</i> drug release	High	<i>In-vitro</i> drug release decreases as the concentration of HPMC E15 increases. Hence, the risk was high.
PEG 4000	Folding endurance	High	Folding endurance decreases as the concentration of PEG 4000 increases as it reduces the brittleness of the MDF. There was direct relationship. Hence, the risk was high.
	Thickness	Low	PEG 4000 was used as a plasticizer. As it doesn't exhibit any major impact on Thickness, the risk was low.
	<i>In-vitro</i> drug release	Medium	<i>In-vitro</i> drug release may vary depend on concentration of PEG 4000. Hence, the risk was medium.
SLS	Folding endurance	Low	Almost negligible increase in Folding endurance as the concentration of SLS increases. Hence, the risk was low.
	Thickness	Low	No major impact was observed on Thickness, so the risk was low.
	<i>In-vitro</i> drug release	Medium	<i>In-vitro</i> drug release may vary depend on concentration of SLS. Hence, the risk was medium.

3.3 Optimization of MDF Formulation

3.3.1 Selection of Experimental Design

Box Behnken design was selected for optimization as it provides information exclusively on the effect of experiment variables and overall experimental error in a minimum number of runs. Being a response surface design, it requires three levels (-1, 0, 1) and offers a total of 15 experimental runs to investigate the impact of independent variables on response variables.

3.3.2 Independent Variables

Table 7: Independent Variables with Levels

Independent Variables	Unit	Low (-1)	High (+1)
Polymer (HPMC E15)	mg	320	400
Plasticizer (PEG 4000)	mg	210	290
Surfactant (SLS)	mg	90	170

3.3.3. Response Variables

Table 8: Response Variables

Response Variables	Unit
Thickness	μm
Folding endurance	folds
% Drug releases	%

3.3.4 Fabrication and Characterization of MDF Batches Suggested by BBD

Table 9: Different Combinations of MDF Formulations Offered by BBD

Run	Batch No.	Independent Variables		
		X1	X2	X3
		HPMC E15 (mg)	PEG 4000 (mg)	SLS (mg)
1.	MQA301	360	250	130
2.	MQA302	320	250	90

3.	MQA303	360	210	90
4.	MQA304	400	250	90
5.	MQA305	400	290	130
6.	MQA306	320	290	130
7.	MQA307	320	250	170
8.	MQA308	400	250	170
9.	MQA309	360	290	170
10.	MQA310	360	250	130
11.	MQA311	360	210	170
12.	MQA312	400	210	130
13.	MQA313	360	290	90
14.	MQA314	320	210	130
15.	MQA315	360	250	130

Table 10: Results of Dependent Variables Analysis of MDF Batches

Run	Batch No.	Response Variables		
		R1	R2	R3
		Thickness (μm)	Folding endurance	Drug release (%)
1.	MQA301	102 ± 0.6324	208 ± 2.9439	91.706
2.	MQA302	88 ± 1.0954	188 ± 0.8165	93.608
3.	MQA303	98 ± 0.8944	198 ± 1.6329	89.176
4.	MQA304	115 ± 1.0954	244 ± 2.1602	85.508
5.	MQA305	120 ± 0.6324	255 ± 2.1602	88.276
6.	MQA306	95 ± 3.1622	191 ± 1.6329	96.406
7.	MQA307	94 ± 1.2649	196 ± 1.6329	96.368
8.	MQA308	118 ± 0.6324	248 ± 2.1602	88.162
9.	MQA309	109 ± 1.4142	236 ± 2.9439	93.12
10.	MQA310	105 ± 0.6324	216 ± 2.9439	91.628
11.	MQA311	101 ± 1.4142	204 ± 3.7416	90.968
12.	MQA312	112 ± 1.0954	246 ± 1.6329	85.084
13.	MQA313	107 ± 0.6324	231 ± 1.6329	92.624
14.	MQA314	88 ± 1.6733	185 ± 0.8165	93.436
15.	MQA315	104 ± 0.6324	224 ± 1.6329	92.048

3.4 Statistical Analysis of Findings of Response Variables for MDF Batches**3.4.1. Fit Summary of MDF Batches****Table 11: Lack of Fit for Thickness**

Source	Sequential P-value	Lack of Fit P-value	Adjusted R^2	Predicted R^2	Remark
Linear	<0.0001	0.9306	0.9947	0.9924	Suggested
2FI	0.4594	0.9395	0.9936	0.9865	-
Quadratic	0.9997	0.7552	0.9948	0.9744	-
Cubic	0.7552	-	0.9979	-	Aliased

Table 12: Lack of Fit for Folding Endurance

Source	Sequential P-value	Lack of Fit P-value	Adjusted R^2	Predicted R^2	Remark
Linear	<0.0001	0.7294	0.9208	0.8838	Suggested
2FI	0.9907	0.5871	0.8925	0.7445	-
Quadratic	0.9649	0.3743	0.8364	0.2809	-
Cubic	0.3743	-	0.8902	-	Aliased

Table 13: Lack of Fit for % Drug Release

Source	Sequential P-value	Lack of Fit P-value	Adjusted R ²	Predicted R ²	Remark
Linear	<0.0001	0.1042	0.9679	0.9528	Suggested
2FI	0.8169	0.0787	0.964	0.984	-
Quadratic	0.1456	0.1072	0.9766	0.8745	-
Cubic	0.1072	-	0.9957	-	Aliased

Table 14: Sequential Model Sum of Square Tables for Thickness

Source	Sum of Square	df	Mean Square	F-value	p-value	Remark
Mean vs Total	1.614E+05	1	1.614E+05	-	-	-
Linear vs Mean	1402.50	3	467.50	492.89	<0.0001	Suggested
2FI vs Linear	2.75	3	0.9167	0.9544	0.4594	-
Quadratic vs 2FI	0.0167	3	0.0056	0.0036	0.9997	-
Cubic vs Quadratic	3.00	3	1.0000	0.4286	0.7552	Aliased
Residual	4.67	2	2.33	-	-	-
Total	1.628E+05	15	10854.80	-	-	-

Table 15: Sequential Model Sum of Square Tables for Folding Endurance

Source	Sum of Square	df	Mean Square	F-value	p-value	Remark
Mean vs Total	7.129E+05	1	7.129E+05	-	-	-
Linear vs Mean	7652.25	3	2550.75	55.26	<0.0001	Suggested
2FI vs Linear	6.50	3	2.17	0.0346	0.9907	-
Quadratic vs 2FI	24.50	3	8.17	0.0856	0.9649	-
Cubic vs Quadratic	384.75	3	116.25	1.82	0.3743	Aliased
Residual	128.00	2	64.00	-	-	-
Total	7.210E+05	15	48068.00	-	-	-

Table 16: Sequential Model Sum of Square Tables for % Drug Release

Source	Sum of Square	df	Mean Square	F-value	p-value	Remark
Mean vs Total	1.248E+05	1	1.165E+05	-	-	-
Linear vs Mean	159.06	3	53.02	141.58	<0.0001	Suggested
2FI vs Linear	0.4306	3	0.1435	0.3113	0.8169	-
Quadratic vs 2FI	2.32	3	0.7743	2.83	0.1456	-
Cubic vs Quadratic	1.27	3	0.4222	8.49	0.1072	Aliased
Residual	0.0995	2	0.0497	-	-	-
Total	1.249E+05	15	8328.68	-	-	-

3.4.2 ANOVA Studies

Table 17: ANOVA for Thickness

Source	Sum of Square	df	Mean Square	F-value	p- value	Remark
Model	1402.50	3	467.50	492.89	<0.0001	significant
A-HPMC E15	1250.00	1	1250.00	1317.89	<0.0001	-
B-PEG 4000	128.00	1	128.00	134.95	<0.0001	-
C-SLS	24.50	1	24.50	25.83	0.0004	-
Residual	10.43	11	0.9485	-	-	-
Lack of fit	5.77	9	0.6407	0.2746	0.9306	not significant
Pure Error	4.67	2	2.33	-	-	-
Cor Total	1412.93	14	-	-	-	-

Table 18: Fit Statistics for Thickness

Particulars	Observations	Particulars	Observations
Std. Dev.	0.9739	R²	0.9926
Mean	103.73	Adjusted R²	0.9906
C.V.%	0.9389	Predicted R²	0.9875
-	-	Adeq Precision	65.6167

3.4.3 ANOVA for Folding Endurance

Table 19: ANOVA for Folding Endurance

Source	Sum of Square	df	Mean Square	F-value	p- value	Remark
Model	7652.25	3	2550.75	55.26	<0.0001	significant
A-HPMC E15	6786.12	1	6786.12	147.02	<0.0001	-
B-PEG 4000	800.00	1	800.00	17.33	0.0016	-
C-SLS	66.13	1	66.13	1.43	0.2565	-
Residual	507.75	11	46.16	-	-	-
Lack of fit	379.75	9	42.19	0.6593	0.7294	not significant
Pure Error	128.00	2	64.00	-	-	-
Cor Total	8160.00	14	-	-	-	-

Table 20: Fit Statistics for Folding Endurance

Particulars	Observations	Particulars	Observations
Std. Dev.	6.79	R²	0.9378
Mean	218.00	Adjusted R²	0.9208
C.V.%	3.12	Predicted R²	0.8838
-	-	Adeq Precision	22.3034

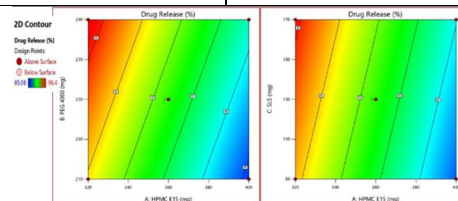
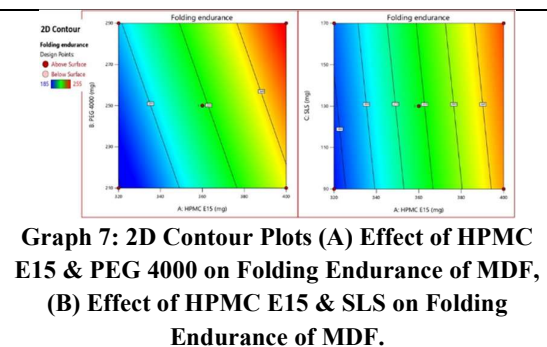
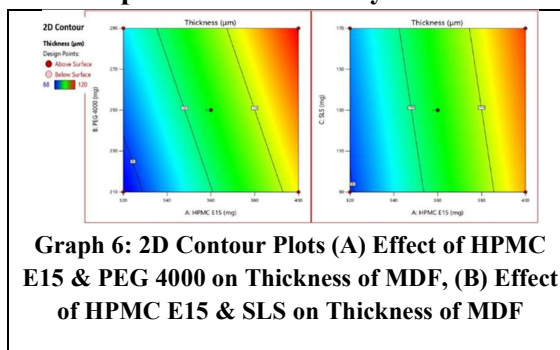
Table 21: Statistical Analysis of % Drug Release

Source	Sum of Square	df	Mean Square	F-value	p- value	Remark
Model	159.06	3	53.02	141.58	<0.0001	significant
A-HPMC E15	134.32	1	134.32	358.56	<0.0001	-
B-PEG 4000	17.32	1	17.32	46.24	<0.0001	-
C-SLS	7.34	1	7.43	19.84	0.0010	-
Residual	4.12	11	0.3745	-	-	-
Lack of fit	4.02	9	0.4467	8.98	0.1042	not significant
Pure Error	0.0995	2	0.0497	-	-	-
Cor Total	163.18	14	-	-	-	-

Table 22: Fit Statistics for % Drug Release

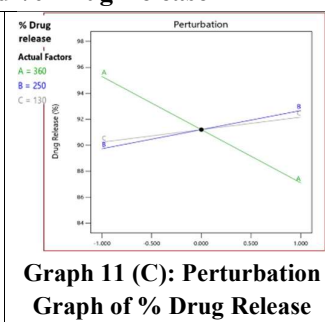
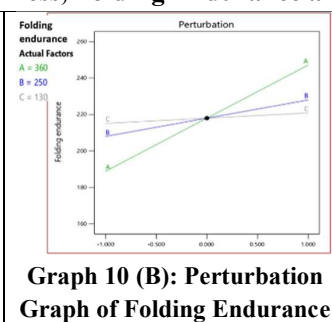
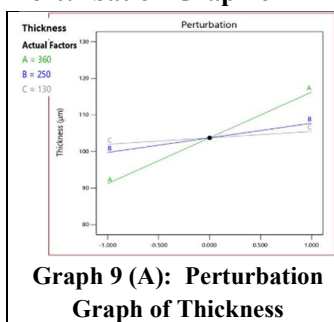
Particulars	Observations	Particulars	Observations
Std. Dev.	0.6120	R ²	0.9748
Mean	91.20	Adjusted R ²	0.9679
C.V.%	0.6710	Predicted R ²	0.9528
-	-	Adeq Precision	35.2435

3.5 Response Surface Analysis of MDF Batches

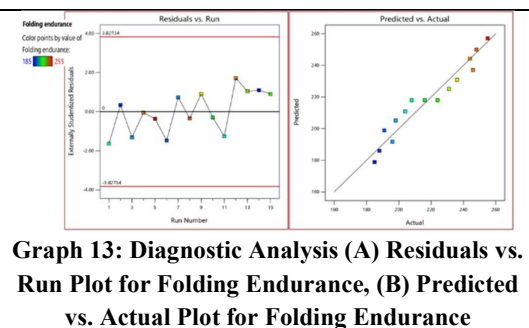
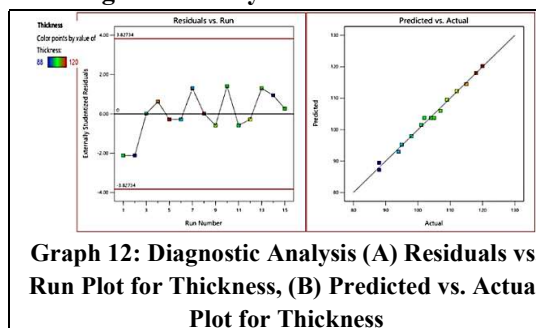


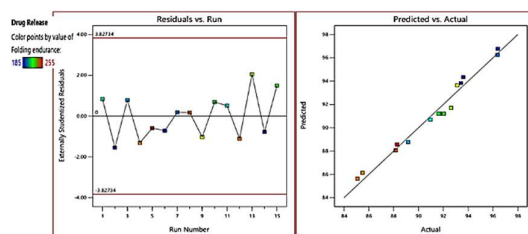
Graph 8: 2D Contour Plots (A) Effect of HPMC E15 & PEG 4000 on % Drug Release of MDF, (B) Effect of HPMC E15 & SLS on % Drug Release of MDF

3.5.1 Perturbation Graph of Thickness, Folding Endurance and % Drug Release



3.5.2 Diagnostic Analysis for Assessment of Residuals





Graph 14: Diagnostic Analysis (A) Residuals vs. Run Plot for % Drug Release, (B) Predicted vs. Actual Plot for % Drug Release

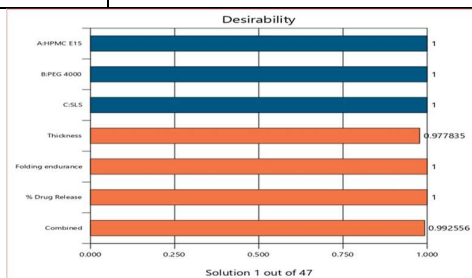
3.6 Prediction of Optimized Labetalol hydrochloride loaded MDF Formulation

Table 23: Desirable Objectives of Response Variables for Optimized MDF Batch

Response variables	Criteria
Thickness	Target
Folding endurance	Target
% drug release	Maximum

Table 24: Software Predicted Values of Responses with Maximum Desirability

Response variables	Predicted values	Desirability
Thickness (μm)	100.44	0.993
Folding endurance (folds)	210	
% drug release (%)	96.36	



Graph 15: The Desirability of Individual Independent and Response Variables Depicted in Bar Graph

3.7 Fabrication of Optimized Labetalol Hydrochloride Loaded MDF Formulation

Table 25: Suggested Formulation Variables of Labetalol Hydrochloride Loaded MDF with Highest Desirability

Variable	Quantity
HPMC E15	331.947 mg
PEG 4000	290.000 mg
SLS	163.748 mg

3.8 Characterization of Optimized Labetalol Hydrochloride Loaded MDF Formulation

3.8.1. Measurement of Thickness, Drug Content Uniformity, pH, Weight Variation, Disintegration Time and Folding Endurance

Table 26: Thickness, Drug Content Uniformity, pH, Weight Variation, Disintegration Time and Folding Endurance of MDF Batches

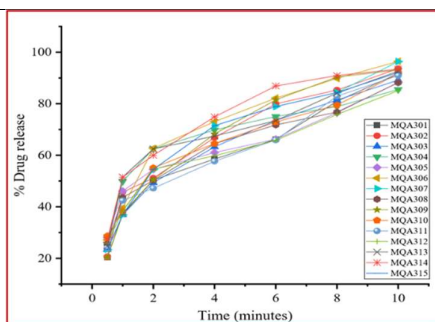
Batch No.	Thickness (μm) (Mean $\pm$ SD)	Folding endurance (folds) (Mean $\pm$ SD)	Drug Content Uniformity (%)	pH (Mean $\pm$ SD)	Weight Variation (mg) (Mean $\pm$ SD)	DT (seconds) (Mean $\pm$ SD)
MQA301	102 $\pm$ 0.632	208 $\pm$ 2.943	96.98	6.67 $\pm$ 0.032	95.50 $\pm$ 0.82	34.6 $\pm$ 1.52
MQA302	88 $\pm$ 1.095	188 $\pm$ 0.816	98.08	6.54 $\pm$ 0.030	91.17 $\pm$ 0.44	36 $\pm$ 1
MQA303	98 $\pm$ 0.894	198 $\pm$ 1.632	94.79	6.81 $\pm$ 0.004	92.86 $\pm$ 0.27	27.6 $\pm$ 0.47
MQA304	115 $\pm$ 1.095	244 $\pm$ 2.160	100	6.68 $\pm$ 0.012	97.15 $\pm$ 0.23	24.6 $\pm$ 1.24

MQA305	120 ± 0.632	255 ± 2.16	96.16	6.71 ± 0.020	105.34 ± 0.26	23 ± 0.81
MQA306	95 ± 3.162	191 ± 1.632	99.45	6.79 ± 0.012	94.5 ± 0.35	28.6 ± 0.94
MQA307	94 ± 1.264	196 ± 1.632	100.2	6.72 ± 0.012	95.89 ± 0.53	29.3 ± 2.05
MQA308	118 ± 0.632	248 ± 2.160	99.17	6.7 ± 0.014	102.44 ± 0.43	33 ± 1.63
MQA309	109 ± 1.414	236 ± 2.943	96.43	6.69 ± 0.008	103.60 ± 0.53	28.6 ± 1.24
MQA310	105 ± 0.632	216 ± 2.943	97.26	6.65 ± 0.016	96.71 ± 0.41	23 ± 0.81
MQA311	101 ± 1.414	204 ± 3.741	98.63	6.78 ± 0.009	96.95 ± 0.08	30.3 ± 1.24
MQA312	112 ± 0.632	246 ± 1.632	93.97	6.62 ± 0.020	97.18 ± 0.18	24.3 ± 0.47
MQA313	107 ± 0.632	231 ± 1.632	99.72	6.66 ± 0.009	95.02 ± 0.20	26.6 ± 0.94
MQA314	88 ± 1.673	185 ± 0.816	95.61	6.83 ± 0.009	91.02 ± 0.08	25.3 ± 0.94
MQA315	104 ± 0.632	224 ± 1.632	94.52	6.64 ± 0.020	95.23 ± 0.32	26 ± 0.81
MQA316	99.2 ± 0.4	209 ± 0.816	99.72	6.77 ± 0.155	104.21 ± 0.58	27.66 ± 1.24

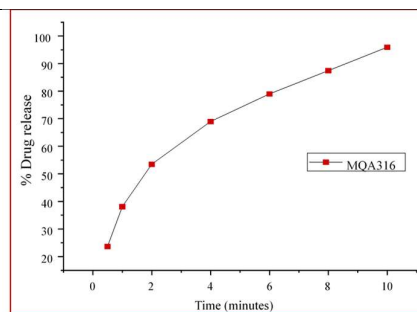
3.8.2. In-vitro Drug Dissolution Study

Table 27: % drug release of MDF Batches

Batches	% Drug Release (%)						
	Time						
	30 sec	1 min	2 min	4 min	6 min	8 min	10 min
MQA301	20.28	37.13	50.39	58.53	66.20	80.94	91.70
MQA302	20.64	45.70	51.14	66.85	79.93	85.29	93.60
MQA303	23.40	36.87	49.35	63.24	73.50	81.40	89.17
MQA304	25.56	49.86	62.45	69.83	75.03	78.87	85.50
MQA305	28.08	46.06	54.50	61.03	66.21	76.51	88.27
MQA306	27.12	39.57	62.66	73.05	82.15	89.82	96.40
MQA307	23.40	36.87	54.51	71.49	78.89	84.35	96.36
MQA308	25.56	43.62	50.22	64.48	71.76	76.75	88.16
MQA309	20.40	37.54	50.52	68.26	81.49	90.36	93.12
MQA310	28.68	38.27	55.10	64.52	72.89	79.45	91.62
MQA311	24	42.40	47.18	57.66	65.91	82.93	90.96
MQA312	25.92	38.23	54.58	59.79	65.79	75.84	85.08
MQA313	24.72	51.53	62.46	67.44	73.45	84.09	92.62
MQA314	27.24	51.45	59.98	74.88	86.89	90.92	93.43
MQA315	23.40	36.87	54.51	71.49	78.89	84.35	92.04
MQA316	23.64	38.07	53.46	68.97	78.97	87.44	95.90



Graph 16: In-vitro % Drug Release Profiles for Batches MQA301 to MQA315



Graph 17: In-vitro % Drug Release Profiles of MQA316.

3.8.3 SEM of Optimized MDF

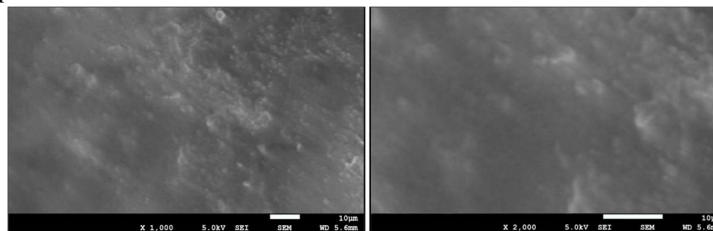
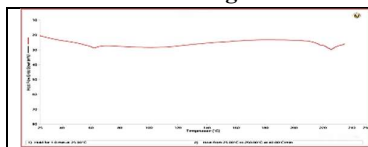


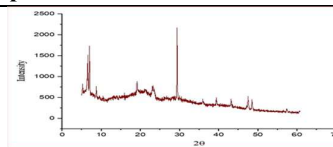
Figure 1: SEM of Labetalol Hydrochloride Loaded Optimized MDF



Graph 18: DSC of Labetalol Hydrochloride Loaded Optimized MDF



Graph 19: FTIR of Labetalol Hydrochloride Loaded Optimized MDF



Graph 20: XRD of Labetalol Hydrochloride Loaded Optimized MDF

3.8.4 Accelerated Stability Study

Table 28: Stability Study Data of Optimized MDF Formulation

Parameters	Time		
	1 Month	2 Month	3 Month
Thickness (µm)	98.6 ± 0.8	98 ± 0.63	97.8 ± 0.4
Disintegration time (sec.)	27 ± 0.81	26.3 ± 0.47	26 ± 0.81
Drug content (%)	99.17	98.9	98.9
Surface pH	6.75 ± 0.07	6.73 ± 0.01	6.72 ± 0.03
Folding endurance (folds)	207 ± 0.47	206 ± 0.81	206 ± 0.47
<i>In-vitro</i> drug release (%)	95.48	94.65	94
Weight variation (mg)	103.71 ± 0.78	103.54 ± 0.66	103.41 ± 0.57

3.9 Validation for Derived Model

Table 29: Summary of the Data of Optimized MDF Batch (MQA316)

Parameters	Observations
Thickness (µm)	99.2 ± 0.4
Disintegration time (sec.)	27 ± 0.81
Drug content (%)	99.17
Surface pH	6.75 ± 0.07
Folding endurance (folds)	209 ± 0.816
<i>In-vitro</i> drug release (%)	95.90
Weight variation (mg)	103.71 ± 0.78
SEM analysis	The Labetalol hydrochloride loaded MDF formulation exhibited a smooth surface with a few small pores (section 6.9.8).
DSC analysis	The DSC thermogram of pure Labetalol hydrochloride had sharp endothermic peaks at 85.06°C and 185°C. While DSC thermogram of optimized Labetalol hydrochloride loaded MDF formulation showed decrease in intensity of peaks confirming its amorphous form (section 6.9.9).
FTIR analysis	The FTIR spectra of pure Labetalol hydrochloride revealed the characteristics absorption bands at 3401.7, 2943.61, 1632.44, 1722.43, 1529.55, 1469.07, 1368.3, 1224.09, and 1104.24. FTIR spectrum of optimized MDF formulation showed remarkable bands with a significant reduction in the intensity of the peaks, indicating there is no unwanted interaction between drug and excipients in the formulation. Hence this confirms the compatibility of the formulation ingredients (section 6.9.10).

XRD analysis	The X-ray diffractogram of Labetalol hydrochloride had shown the characteristic sharp peaks at 9.1187, 16.8147, 18.3423, 21.4748 and 25.2841 on diffraction angle (2 θ) suggesting that the drug is present in a crystalline form. While the XRD pattern of optimized formulation showed few characteristic diffraction peaks with reduced intensity indicating the conversion of drug to amorphous or molecular form (section 6.9.11).
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Table 30: Predicted and Observed Response Variables for Labetalol Hydrochloride Loaded MDF

Response variables	Predicted values	Observed values
Thickness (μm)	100.44	99.2 ± 0.4
Folding endurance (folds)	210	209 ± 0.816
<i>In-vitro</i> drug release (%)	96.36	95.90

Discussion

The present study successfully optimized Labetalol HCl mouth dissolving film using a Quality by Design approach and Box-Behnken Design. HPMC E15, PEG 4000, and SLS significantly influenced the film characteristics, particularly thickness, folding endurance, and drug release. Increasing HPMC E15 increased film thickness and folding endurance but reduced drug release, whereas PEG 4000 and SLS enhanced drug release. The optimized formulation (MQA316) containing HPMC E15 331.947 mg, PEG 4000 290 mg, and SLS 163.748 mg showed satisfactory characteristics, including thickness of $99.2 \pm 0.4 \mu\text{m}$, folding endurance of 209 ± 0.816 folds, drug content of 99.17%, disintegration time of approximately 27 seconds, and *in-vitro* drug release of 95.90%. The experimental results were closely comparable with the predicted values, confirming the suitability of the developed model. SEM showed a smooth film surface with few small pores, while DSC and XRD indicated reduced crystallinity of the drug in the optimized formulation. FTIR studies showed no significant unwanted interaction between the drug and excipients. Furthermore, the formulation remained stable during three months of accelerated stability testing. Overall, the findings demonstrate that QbD and DoE provide an effective systematic approach for optimizing Labetalol HCl mouth dissolving film with desirable quality attributes.

4. CONCLUSION

The MDF of Labetalol hydrochloride was successfully developed and optimized using a computer-assisted QbD approach. The important properties and CQAs were identified, assessed, and controlled in the formulation using a QbD approach. As a result, the predefined QTPP was matched, confirming the formulation's safety and efficacy. The optimized formulation of MDFs of Labetalol HCL, revealed the highest desirability of 0.993 which is close to 1.

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